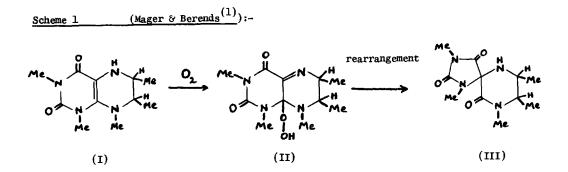
THE AUTOXIDATIVE RING CONTRACTION OF BLOCKED TETRAHYDROPTERIDINES: ORIGIN OF INTRODUCED OXYGEN FUNCTION.

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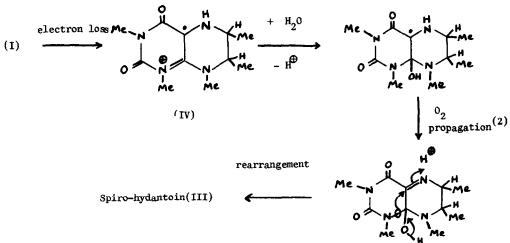
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(Received in UK 15 March 1973; accepted for publication 28 March 1973) It has been established⁽¹⁾ that 1,3,6,7,8 - pentamethyl - 5,6,7,8 - tetrahydropteri -2,4 - dione (I) undergoes a contraction of the pyrimidine ring, when autoxidised in water, to give 4,1',3' - trimethyl - 3 - oxopiperazine - 2 - spiro - 5' - hydantoin (III). This has been presented⁽¹⁾ as evidence for the intermediacy of transient 8a - hydroperoxides(II) in the autoxidation (scheme 1), a postulate which is not in agreement with kinetic data which we have reported⁽²⁾.

We have now shown conclusively that the new oxygen function introduced during the reaction is derived from the water used as solvent. The blocked pteridine (10 mg.) was dissolved in pyridine/water (10:1, 2 cm³) and shaken under oxygen for two days, and the solvent was then removed under vacuum. The mass spectrum of the product corresponded to a mixture of spiro - hydantoin⁽¹⁾ (M^+ , m/e = 254) and starting material (M^+ , m/e = 238). The experiment was repeated using oxygen - 18 enriched water (43.46% enrichment), when the resulting mass spectrum showed that no isotopic exchange between the starting material amide groups and water had occurred $(M^{\dagger} \text{ at } ^{m}/e = 238 \text{ obtained}, \text{ but no peaks at } ^{m}/e 240$ or 242). However, the spiro - hydantoin spectrum showed the incorporation of $\frac{18}{0}$ (M : M + 2 ratio 1.6 ; theoretical ratio, from water = 1.53). Treatment of the labelled product with H_2^{10} did not lead to loss of label (shown by persistence of M + 2 at $\frac{m}{e} = 256$). This demonstrates that the label is introduced by addition of water and not through exchange and this is not consistent with the mechanism proposed by Mager and Berends⁽¹⁾. It is probable that the addition occurs at the radical - cation stage (IV) and not at the quinonoid stage, since no 7,8 - dihydropteridine is observed in the products $^{(1)}$, as would be expected from rearrangement of the quinonoid form (see, for example, reference 2). A reasonable mechanism is presented in scheme 2, not requiring the formation of 8a - hydroperoxides, and in which molecular oxygen is reduced to hydrogen peroxide.



Scheme 2:-



Acknowledgements

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References.

(1)	H.I.X. Mager and W. Berends,	Rec. Trav. Chim. <u>91</u> , 1137, (1972)
(2)	J.A. Blair and A.J. Pearson,	Tetrahedron Lett., No. 3, 203, (1973).